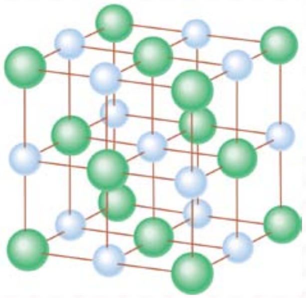


THE BIG IDEA

Solids can be described in terms of crystal structure, density, and elasticity.

Humans have been classifying and using solid materials for many thousands of years. Not until recent times has the discovery of atoms and their interactions made it possible to understand the structure of materials. We have progressed from being finders and assemblers of materials to actual makers of materials.

18.1 Crystal Structure

✓ **The shape of a crystal mirrors the geometric arrangement of atoms within the crystal.**

18.1 Crystal Structure

Minerals such as quartz, mica, or galena have many smooth, flat surfaces at angles to one another.

The minerals are made of **crystals,** or regular geometric shapes whose components are arranged in an orderly, repeating pattern.

The mineral samples themselves may have very irregular shapes, as if they were small units stuck together.

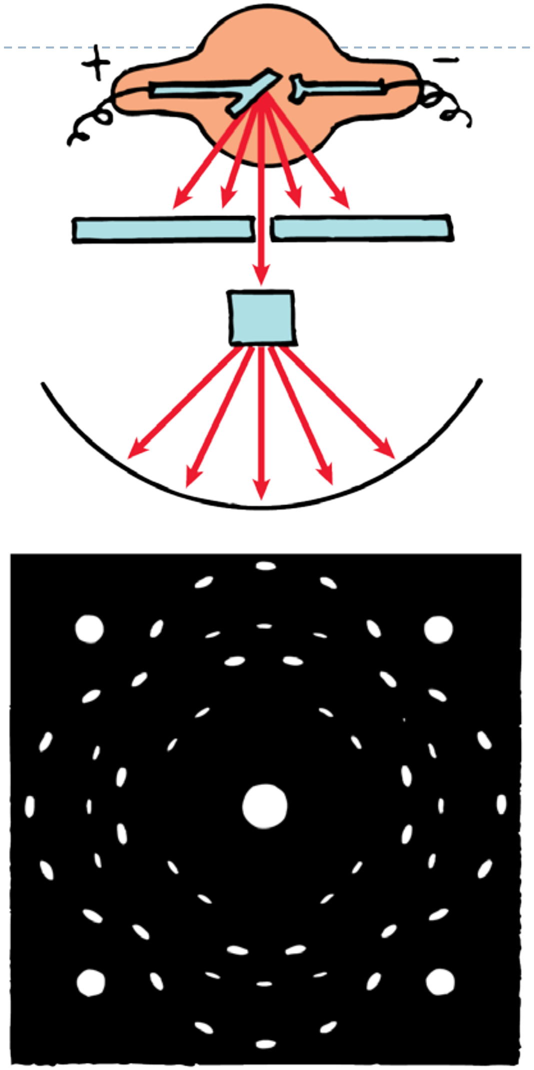
18.1 Crystal Structure

Not all crystals are evident to the naked eye.
Their existence in many solids was not discovered until X-rays became a tool of research early in the twentieth century.

When X-rays pass through a crystal of common table salt (sodium chloride), they produce a distinctive pattern on photographic film.

The patterns made by X-rays on photographic film show that the atoms in a crystal have an orderly arrangement.

Every crystalline structure has its own unique X-ray pattern.



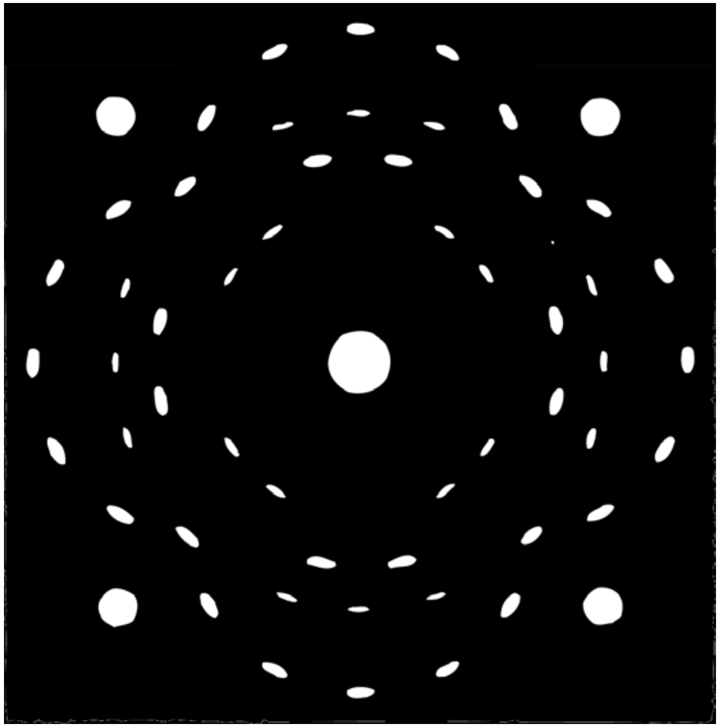
18.1 Crystal Structure

The radiation that penetrates the crystal produces the pattern shown on the photographic film beyond the crystal.

The white spot in the center is caused by the main unscattered beam of X-rays.

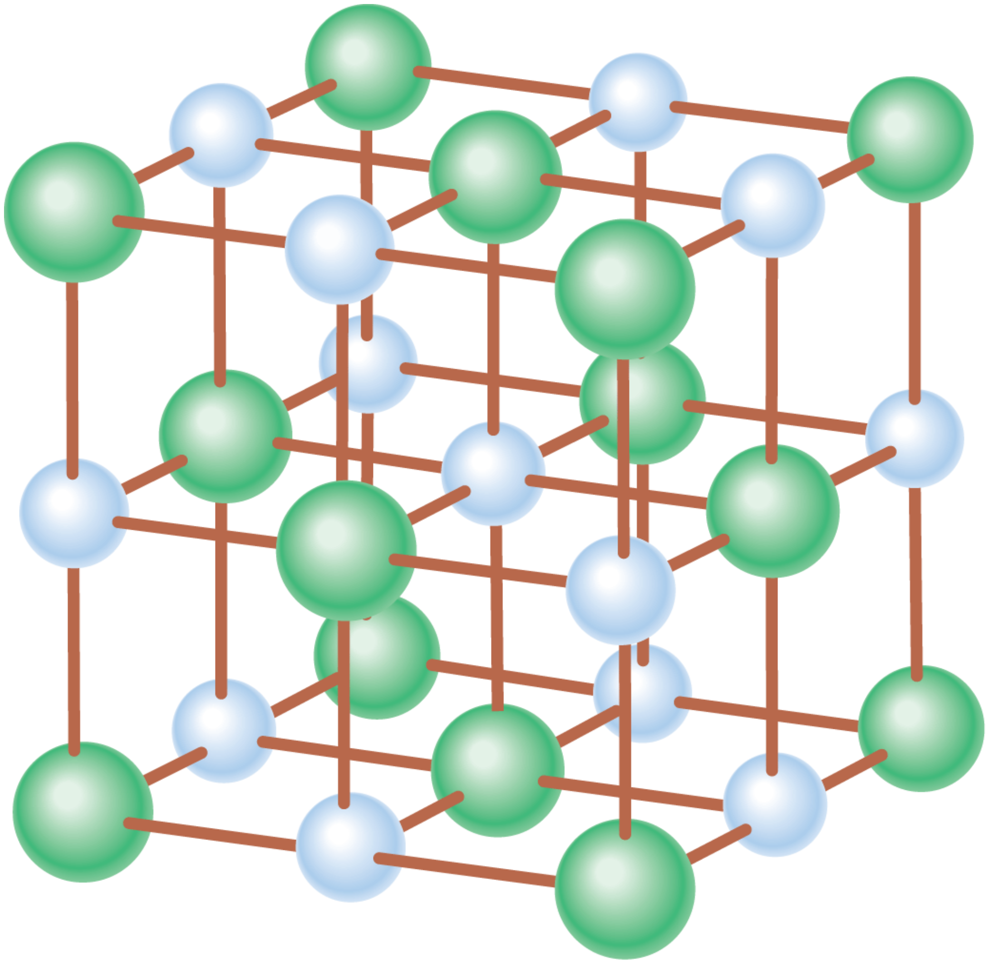
The size and arrangement of the other spots indicate the arrangement of sodium and chlorine atoms in the crystal.

All crystals of sodium chloride produce this same design.



18.1 Crystal Structure

In this model of a sodium chloride crystal, the large spheres represent chloride ions, and the small ones represent sodium ions.



18.1 Crystal Structure

CONCEPT CHECK

What determines the shape of a crystal?

18.2 Density

- **The density of a material depends upon the masses of the individual atoms that make it up, and the spacing between those atoms.**

18.2 Density

One of the properties of solids, as well as liquids and even gases, is the measure of how tightly the material is packed together.

Density is a measure of how much matter occupies a given space; it is the amount of mass per unit volume:

$$\text{density} = \frac{\text{mass}}{\text{volume}}$$

18.2 Density

When the loaf of bread is squeezed, its volume decreases and its density increases.

18.2 Density

Density is a property of a material; it doesn't matter how much you have.

- A pure iron nail has the same density as a pure iron frying pan.
- The pan may have 100 times as many iron atoms and 100 times as much mass, so it will take up 100 times as much space.
- The mass per unit volume for the iron nail and the iron frying pan is the same.

Iridium is the densest substance on Earth.

Individual iridium atoms are less massive than atoms of gold, mercury, lead, or uranium, but the close spacing of iridium atoms in an iridium crystal gives it the greatest density.

A cubic centimeter of iridium contains more atoms than a cubic centimeter of gold or uranium.

18.2 Density

Table 18.1 Densities of a Few Substances

Solids	Density (g/cm^3)	Liquids	Density (g/cm^3)
Iridium	22.6	Mercury	13.6
Osmium	22.6	Glycerin	1.26
Platinum	21.4	Sea water	1.03
Gold	19.3	Water at 4°C	1.00
Uranium	19.0	Benzine	0.90
Lead	11.3	Ethyl alcohol	0.81
Silver	10.5		
Copper	8.9		
Brass	8.6		
Iron	7.8		
Steel	7.8		
Tin	7.3		
Diamond	3.5		
Aluminum	2.7		
Graphite	2.25		
Ice	0.92		
Pine wood	0.50		
Balsa wood	0.12		

18.2 Density

Density varies somewhat with temperature and pressure, so, except for water, densities are given at 0°C and atmospheric pressure.

Water at 4°C has a density of 1.00 g/cm^3.

The gram was originally defined as the mass of a cubic centimeter of water at a temperature of 4°C.

A gold brick, with a density of 19.3 g/cm^3, is 19.3 times more massive than an equal volume of water.

18.2 Density

think!

Which has greater density—1 kg of water or 10 kg of water? 5 kg of lead or 10 kg of aluminum?

Answer:

The density of *any* amount of water (at 4°C) is 1.00 g/cm^3. Any amount of lead always has a greater density than any amount of aluminum.

18.2 Density

CONCEPT CHECK

What determines the density of a material?

18.3 Elasticity

A body's elasticity describes how much it changes shape when a deforming force acts on it, and how well it returns to its original shape when the deforming force is removed.

18.3 Elasticity

Hang a weight on a spring and the spring stretches. Add additional weights and the spring stretches still more.

Remove the weights and the spring returns to its original length.

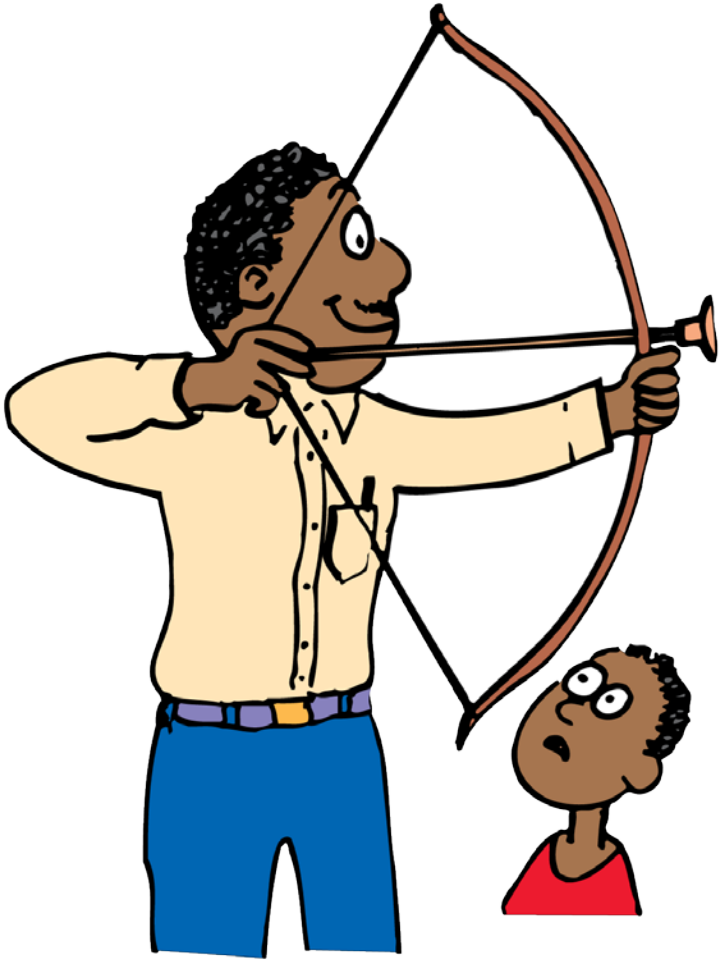
A material that returns to its original shape after it has been stretched or compressed is said to be **elastic.**

18.3 Elasticity

When a bat hits a baseball, it temporarily changes the ball's shape.

When an archer shoots an arrow, he first bends the bow, which springs back to its original form when the arrow is released.

The spring, the baseball, and the bow are elastic objects.



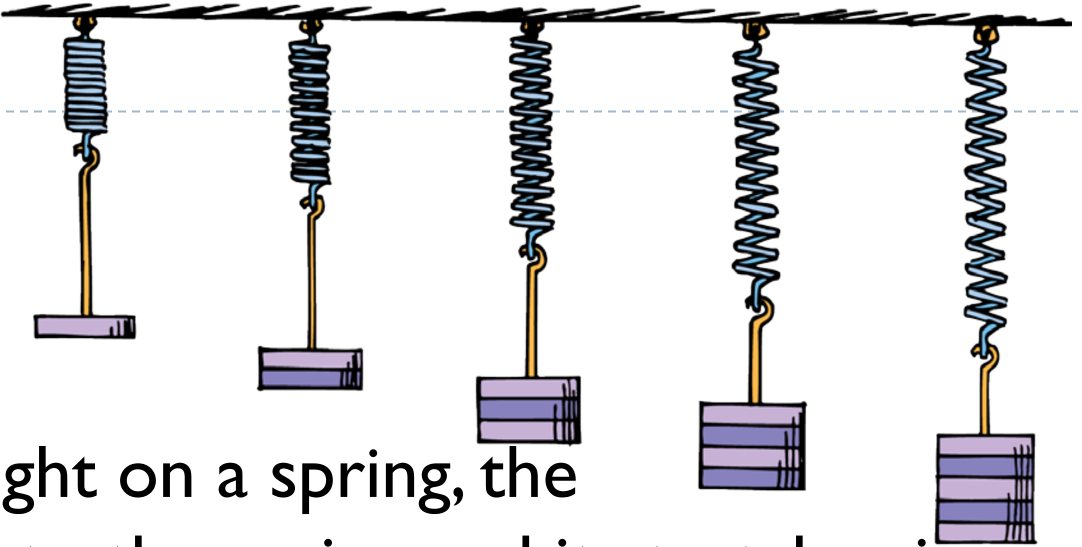
18.3 Elasticity

Not all materials return to their original shape when a deforming force is applied and then removed.

Materials that do not resume their original shape after being distorted are said to be **inelastic.**

Clay, putty, and dough are inelastic materials. Lead is also inelastic, since it is easy to distort it permanently.

18.3 Elasticity



When you hang a weight on a spring, the weight applies a force to the spring and it stretches in direct proportion to the applied force.

According to **Hooke's law,** the amount of stretch (or compression), *x*, is directly proportional to the applied force *F*.

Double the force and you double the stretch; triple the force and you get three times the stretch, and so on: $F \sim \Delta x$

18.3 Elasticity

If an elastic material is stretched or compressed more than a certain amount, it will not return to its original state.

The distance at which permanent distortion occurs is called the **elastic limit.**

Hooke's law holds only as long as the force does not stretch or compress the material beyond its elastic limit.

18.3 Elasticity

think!

A tree branch is found to obey Hooke’s law. When a 20-kg load is hung from the end of it, the branch sags 10 cm. If a 40-kg load is hung from the same place, how much will the branch sag? What would you find if a 60-kg load were hung from the same place? (Assume none of these loads makes the branch sag beyond its elastic limit.)

18.3 Elasticity

think!

A tree branch is found to obey Hooke's law. When a 20-kg load is hung from the end of it, the branch sags 10 cm. If a 40-kg load is hung from the same place, how much will the branch sag? What would you find if a 60-kg load were hung from the same place? (Assume none of these loads makes the branch sag beyond its elastic limit.)

Answer:

A 40-kg load has twice the weight of a 20-kg load. In accord with Hooke's law, $F \sim \Delta x$, the branch should sag 20 cm. The weight of the 60-kg load will make the branch sag 30 cm.

18.3 Elasticity

think!

If a force of 10 N stretches a certain spring 4 cm, how much stretch will occur for an applied force of 15 N?

18.3 Elasticity

think!

If a force of 10 N stretches a certain spring 4 cm, how much stretch will occur for an applied force of 15 N?

Answer:

The spring will stretch 6 cm. By ratio and proportion:

$$\frac{10\ \text{N}}{4\ \text{cm}} = \frac{15\ \text{N}}{x}$$

Then $x = (15\ \text{N}) \times (4\ \text{cm})/(10\ \text{N}) = 6$ cm.

18.3 Elasticity

CONCEPT CHECK

What characteristics are described by an object's elasticity?

18.4 Compression and Tension

- A horizontal beam supported at one or both ends is under stress from the load it supports, including its own weight. It undergoes a stress of both compression and tension (stretching).

18.4 Compression and Tension

Steel is an excellent elastic material. It can be stretched and it can be compressed.

Because of its strength and elastic properties, steel is used to make not only springs but also construction girders.

Vertical steel girders undergo only slight compression.

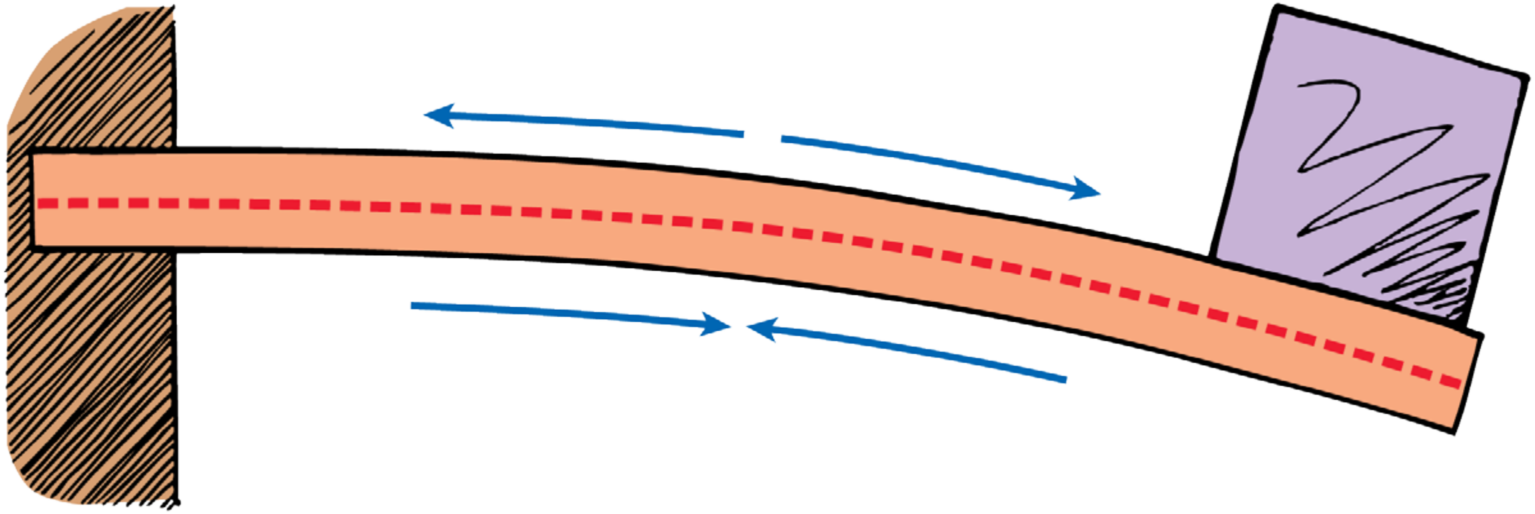
A 25-meter-long vertical girder is compressed about a millimeter when it carries a 10-ton load.

Most deformation occurs when girders are used horizontally, where the tendency is to sag under heavy loads.

The girder sags because of its own weight and because of the load it carries at its end.

18.4 Compression and Tension

The top part of the beam is stretched and the bottom part is compressed. The middle portion is neither stretched nor compressed. (Note that a beam in this position is known as a *cantilever beam*.)



18.4 Compression and Tension

Neutral Layer

The top part of the horizontal beam is stretched. Atoms are tugged away from one another and the top part is slightly longer.

The bottom part of the beam is compressed. Atoms there are pushed toward one another, so the bottom part is slightly shorter.

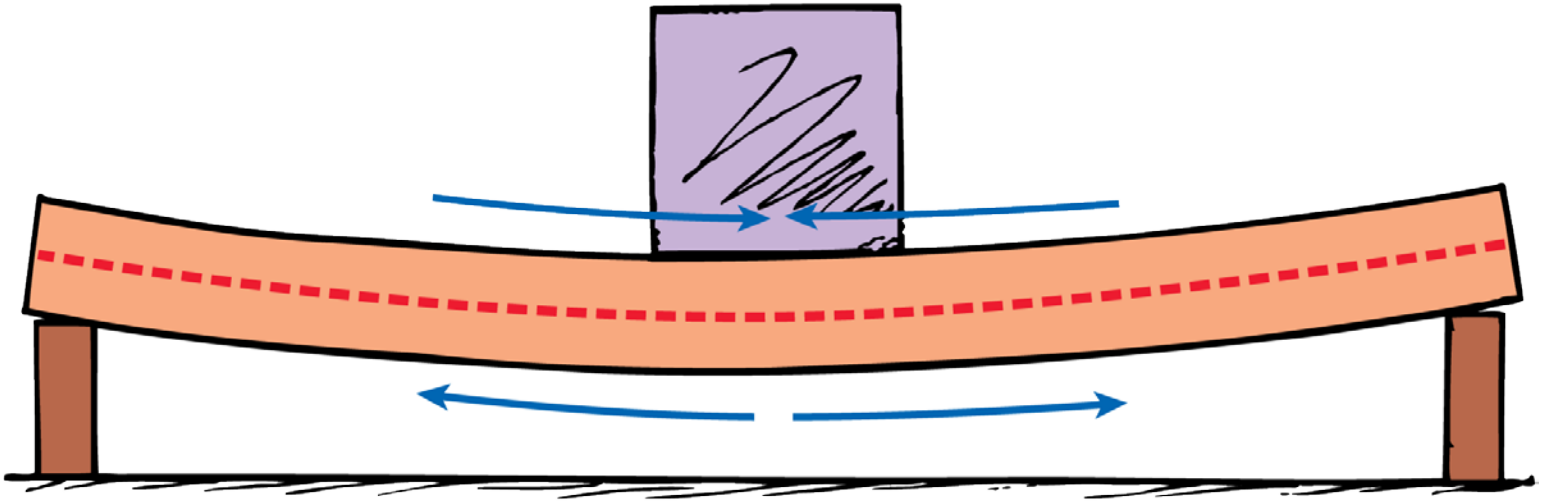
Between the top and bottom, there is a region that is neither stretched nor compressed. This is the *neutral layer*.

18.4 Compression and Tension

Consider a beam that is supported at both ends, and carries a load in the middle.

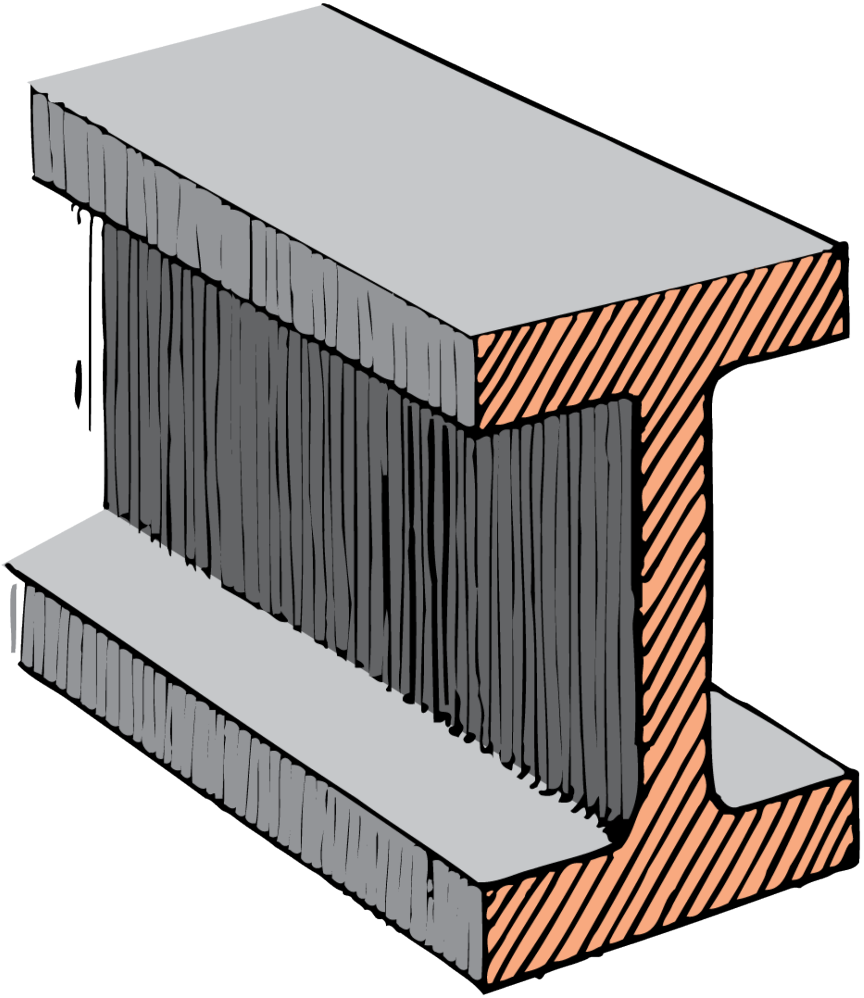
The top of the beam is in compression and the bottom is in tension.

Again, there is a neutral layer along the middle portion of the length of the beam.



18.4 Compression and Tension

An I-beam is like a solid bar with some of the steel scooped from its middle where it is needed least. The beam is therefore lighter for nearly the same strength.



18.4 Compression and Tension

The stress is predominantly in the top and bottom flanges when the beam is used horizontally in construction.

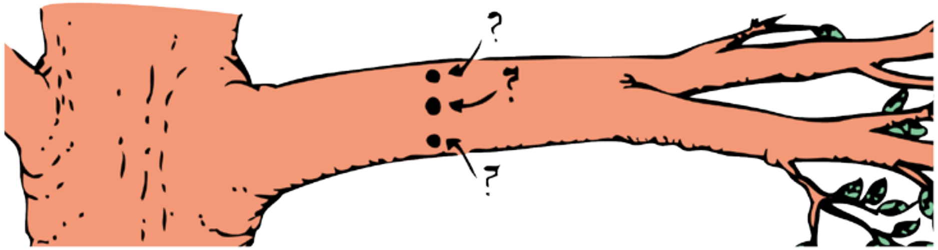
One flange is stretched while the other is compressed. The web is a region of low stress that holds the top and bottom flanges apart.

Heavier loads are supported by farther-apart flanges.

18.4 Compression and Tension

think!

If you make a hole horizontally through the tree branch, what location weakens it the least—the top, the middle, or the bottom?



18.4 Compression and Tension

think!

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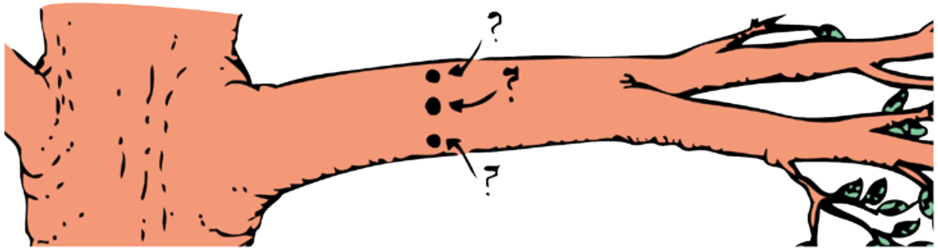


Answer:

The middle. Fibers in the top part of the branch are stretched and fibers in the lower part are compressed. In the neutral layer, the hole will not affect the strength of the branch.

18.4 Compression and Tension

CONCEPT CHECK

How is a horizontal beam affected by the load it supports?

18.5 Scaling

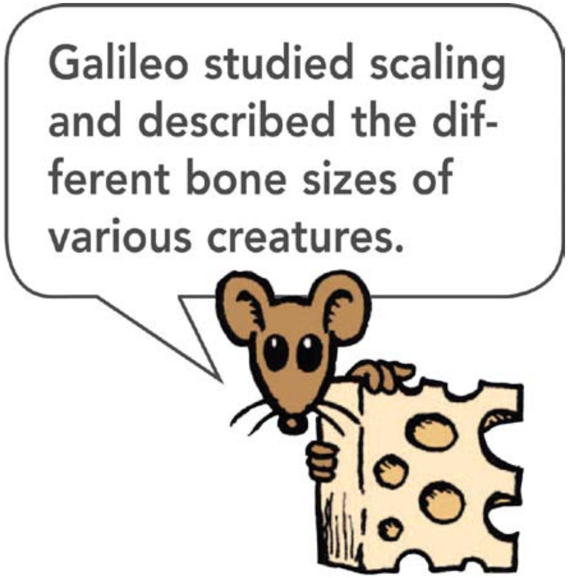
When linear dimensions are enlarged, the cross-sectional area (as well as the total surface area) grows as the square of the enlargement, whereas volume and weight grow as the cube of the enlargement. As the linear size of an object increases, the volume grows faster than the total surface area.

18.5 Scaling

The proportions of things in nature are in accord with their size.

The study of how size affects the relationship between weight, strength, and surface area is known as **scaling.**

As the size of a thing increases, it grows heavier much faster than it grows stronger.



18.5 Scaling

Consider an athlete who can lift his weight with one arm.

- Scaled up to twice his size, every linear dimension would be enlarged by a factor of 2.
- His twice-as-thick arms would have four times the cross-sectional area, so he would be four times as strong.
- His volume would be eight times as great, so he would be eight times as heavy.
- For comparable effort, he could lift only half his weight. *In relation to his weight*, he would be weaker than before.

18.5 Scaling

How Scaling Affects Surface Area vs. Volume

How does surface area compare with volume?

Volume grows as the cube of the enlargement, and both cross-sectional area and total surface area grow as the square of the enlargement.

As an object grows, the ratio of surface area to volume *decreases*.

18.5 Scaling

Smaller objects have more surface area per kilogram. Cooling occurs at the surfaces of objects, so crushed ice will cool a drink faster than an ice cube of the same mass. Crushed ice presents more surface area to the beverage.

18.5 Scaling

The rusting of iron is also a surface phenomenon.
The greater the amount of surface exposed to the air, the faster rusting takes place.
Small filings and "steel wool" are soon eaten away. The same mass of iron in a solid cube or sphere rusts very little in comparison.

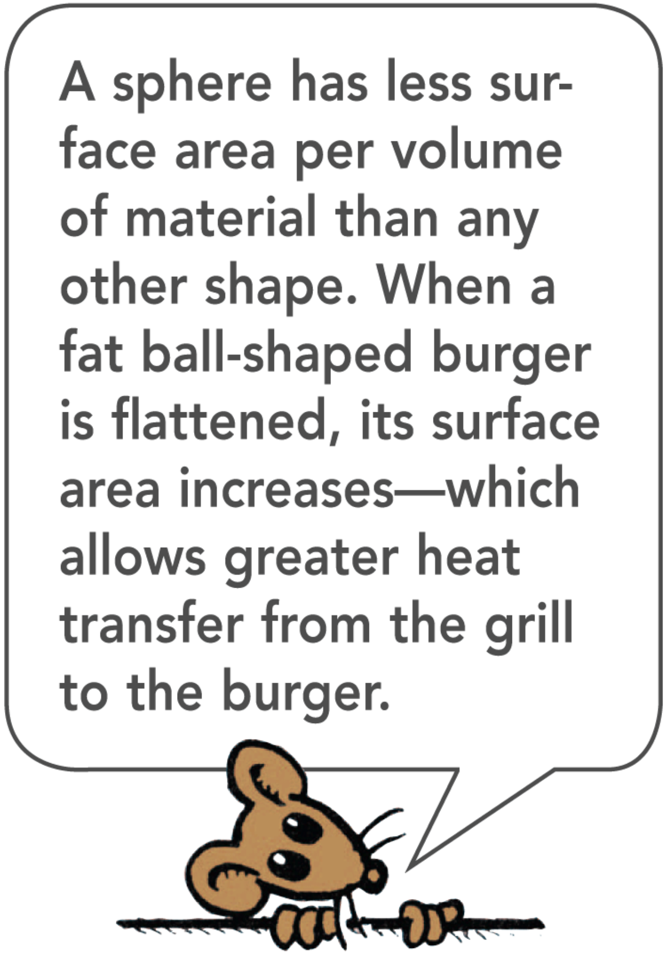
18.5 Scaling

- Chunks of coal burn, while coal dust explodes when ignited.

- Thin French fries cook faster in oil than fat fries.

- Flat hamburgers cook faster than meatballs of the same mass.

- Large raindrops fall faster than small raindrops.



A sphere has less surface area per volume of material than any other shape. When a fat ball-shaped burger is flattened, its surface area increases—which allows greater heat transfer from the grill to the burger.